

## Crystal structure refinements and magnetic behavior of $U_6Ni$ , $UNi_5$ , $UNi_2$ and the substitution derivative $UNi_{1.7}Si_{0.3}$

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The crystal structures of the binaries  $U_6Ni$ ,  $UNi_2$  and  $UNi_5$  have been refined for the first time from single crystal data.  $U_6Ni$  crystallizes in the tetragonal system,  $U_6Mn$  type, s.g.  $I4/mcm$ , with lattice parameters  $a = 10.3835(8)$  Å,  $c = 5.1565(4)$  Å. The uranium atoms occupy  $8h$  (U1) and  $16k$  (U2) Wyckoff positions, and the Ni atoms a  $4a$  position.  $UNi_5$  crystallizes in the cubic  $AuBe_5$  type, s.g.  $F\bar{4}3m$ , with  $a = 6.7958(6)$  Å and the following atom site occupations: U in  $4a$ , Ni1 in  $16e$  and Ni2 in  $4c$ .  $UNi_2$  crystallizes in the hexagonal  $MgZn_2$  type, s.g.  $P6_3/mmc$ , with  $a = 4.9701(4)$  Å,  $c = 8.2527(8)$  Å. Uranium occupies a  $4f$  position and the nickel atoms are in  $2a$  (Ni1) and  $6h$  (Ni2) positions. The shortest interatomic U-U distances are 2.619(2) Å in  $U_6Ni$ , 3.026(1) Å in  $UNi_2$ , and 4.805(1) Å in  $UNi_5$ .  $UNi_2$  shows itinerant ferromagnetism below 25 K, which vanishes with substitution of silicon for nickel.

Uranium compounds / Intermetallics / Crystal structure / Magnetic properties

### 1. Introduction

The crystal structure types of the binary uranium-nickel intermetallics  $U_6Ni$ ,  $UNi_2$  and  $UNi_5$  are known, but to our knowledge their crystal structures had not been previously refined from single crystal data. This work is part of our program concerning a more detailed characterization of the binary uranium intermetallics. We have recently shown by single crystal X-ray diffraction [1-3] that the uranium-nickel compounds formerly reported as  $U_7Ni_9$  and  $U_5Ni_7$  [4] in fact correspond to  $U_{10}Ni_{13}$  and  $U_{11}Ni_{16}$  with original structure types, and that the phases denoted  $\epsilon$ - $\delta$  corresponds to  $UNi_{5-x}$  ( $0.9 < x < 1$ ) with the hexagonal  $CaCu_5$  type.

### 2. Experimental

The elements used were depleted uranium plates (99.8% purity), cleaned in diluted  $HNO_3$  before use), and nickel wire (99.5% purity). Samples with nominal compositions were prepared by direct melting of the two elements in an arc-furnace, under partial pressure of purified argon atmosphere. The buttons were inverted and remelted two times to ensure complete homogenization. Single crystals suitable for X-ray data collection were obtained by annealing  $U_6Ni$  and  $UNi_2$  in sealed quartz tubes for

two weeks at 790°C and 820°C, respectively.

The single crystal X-ray diffraction data were collected on a Nonius Kappa CCD diffractometer. The unit-cell parameters and the orientation matrix were initially determined from the first ten measured frames of the data, and refined during the indexing and the intensity integration process of all the recorded images, using the program DENZO of the Kappa CCD software package [5]. All structure refinements were carried out using SHELXL-97 [6]. Crystal data and structure refinement parameters for all the studied compounds are gathered in Table 1. Magnetization measurements were performed using a SHE Squid magnetometer in the temperature and field ranges 2-300 K and 0-3 T.

### 3. Results

#### 3.1. $U_6Ni$

Single crystals suitable for X-ray data collection were obtained by annealing a sample of  $U_6Ni$  in a sealed quartz tube for two weeks at 790°C, which is the reported peritectic decomposition temperature [4].  $U_6Ni$  crystallizes in the tetragonal system,  $U_6Mn$  type, s.g.  $I4/mcm$ , with lattice parameters  $a = 10.3835(8)$  Å,  $c = 5.1565(4)$  Å. The uranium atoms occupy  $8h$  (U1) and  $16k$  (U2) Wyckoff positions, and the Ni atoms a  $4a$  position (Table 2a). Interatomic

**Table 1** Crystal data and structure refinement parameters.

	$U_6Ni$	$UNi_5$	$UNi_2$	$UNi_{1.7}Si_{0.3}$
Space group	$I4/mcm$ (No 140)	$F\bar{4}3m$ (No 216)	$P6_3/mmc$ (No 194)	$P6_3/mmc$ (No 194)
Lattice parameters (Å)	$a = 10.3835(8)$ $c = 5.1565(3)$	$a = 6.7958(6)$	$a = 4.9701(4)$ $c = 8.2527(8)$	$a = 5.1312(3)$ $c = 7.8207(6)$
Cell volume (Å <sup>3</sup> )	555.96	313.85	176.55(3)	178.33(2)
Formula units per cell	Z = 4	Z = 4	Z = 4	Z = 4
Formula weight (g)	1486.9	531.6	355.4	346.2
Density (calculated) (Mg/m <sup>3</sup> )	17.76	11.25	13.37	12.89
Crystal size (mm <sup>3</sup> )	$0.04 \times 0.04 \times 0.05$	$0.05 \times 0.05 \times 0.05$	$0.04 \times 0.06 \times 0.04$	$0.05 \times 0.06 \times 0.06$
Wavelength (Å)	0.71073	0.71073	0.71073	0.71073
Scan range (deg.)	$3 < \theta < 35^\circ$	$5 < \theta < 35^\circ$	$5 < \theta < 35^\circ$	$5 < \theta < 35^\circ$
Range in $hkl$	$-11 \leq h \leq 16$ $-8 \leq k \leq 11$ $-8 \leq l \leq 8$	$-10 \leq h \leq 10$ $-10 \leq k \leq 10$ $-10 \leq l \leq 8$	$-7 \leq h \leq 8$ $-7 \leq k \leq 8$ $-12 \leq l \leq 13$	$0 \leq h \leq 8$ $-7 \leq k \leq 0$ $0 \leq l \leq 12$
Linear absorption coefficient (mm <sup>-1</sup> )	177.26	80.48	112.02	123.14
Total number of reflections	1161	1429	2607	263
Independent reflections	359	94	173	179
Reflections observed (>2 $\sigma$ )	268	93	149	158
Goodness-of-fit	0.874	1.231	1.069	1.191
Conventional residual (F)	$R_f = 0.0389$	$R_f = 0.0194$	$R_f = 0.0340$	$R_f = 0.0396$
Weighted residual, $w = 1/\sigma^2$ (F)	$R_w = 0.0946$	$R_w = 0.0413$	$R_w = 0.0838$	$R_w = 0.1001$

U-U distances (Table 3a) are as short as those in uranium metal: U2-U2 = 2.619(2) Å, U1-U1 = 2.821(3) Å, U1-U2 = 2.872(7) Å, leading to a nearly temperature-independent paramagnetism in the range 2-300 K. The value of the magnetic susceptibility is  $29 \cdot 10^{-4}$  emu/mole at 300 K. This compound was reported to exhibit superconductivity below  $T_c = 0.33$  K [7].

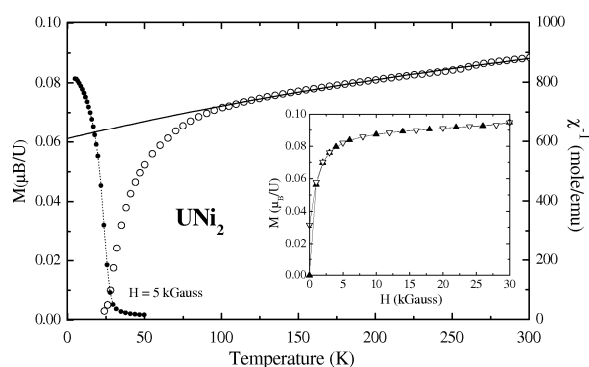
### 3.2. $UNi_5$

Single crystals of  $UNi_5$  could be obtained by annealing an arc-melted sample at 1300°C for 5 hours under vacuum in a high-frequency furnace.  $UNi_5$  crystallizes in the cubic system, AuBe<sub>5</sub> type, s.g.  $F\bar{4}3m$ , with  $a = 6.7958(6)$  Å and atom site occupations: U in 4a, Ni1 in 16e and Ni2 in 4c (Table 2b). The shortest interatomic distances (Table 3b) are U-U = 4.805(1) Å, and U-Ni1 = 2.820(1) Å.  $UNi_5$  is a quasi temperature-independent paramagnet from 2 K to 300 K with a value of the magnetic susceptibility of  $25 \cdot 10^{-3}$  emu/mole at 300 K. In  $UNi_5$ , the 5f delocalization results from the U(5f)-Ni(3d) hybridization.

### 3.3. $UNi_2$ and $UNi_{1.7}Si_{0.3}$

Single crystals of  $UNi_2$  were obtained after annealing an arc-melted stoichiometric sample at 820°C for one week. Refinement of the crystal structure confirmed that  $UNi_2$  crystallizes with the hexagonal MgZn<sub>2</sub> type (s.g.  $P6_3/mmc$ ), whereas the other Laves phases  $UM_2$  with M = Mn, Fe, Co crystallize in the cubic MgCu<sub>2</sub> type. The lattice parameters are:  $a = 4.9701(4)$  Å,  $c = 8.2527(8)$  Å. Uranium occupies a 4f position and the nickel atoms are in 2a (Ni1) and 6h (Ni2) positions (Table 2c). The shortest interatomic distances are U-

U = 3.026(1) Å and U-Ni1 = 2.909(1) Å (Table 3c). In agreement with previous reports [8], our magnetization measurements on polycrystalline annealed samples indicated weak ferromagnetism, with a Curie temperature  $T_C = 25$  K (Fig. 1). The value of the magnetization is  $0.095 \mu_B$  at 5 K under 3 T. A rather large anisotropy of the magnetization was revealed by magnetization on single crystal [9]. Considering that the U-U distances are far below the Hill limit (3.40 Å) implying a very large 5f electron delocalization, the unusual magnetic correlations observed in  $UNi_2$  have been interpreted in terms of itinerant ferromagnetism, associated with a large orbital moment of the 5f electrons [10].



**Fig. 1** Thermal variation of the magnetization (filled circles) and of the reverse susceptibility (empty circles) under 5 kGauss for  $UNi_2$ . Inset: Magnetization as a function of the applied field at  $T = 5$  K. Filled triangles = field up; empty triangles = field down.

**Table 2**

(a) Atomic position parameters for U<sub>6</sub>Ni

Atom	Site	x	y	z	U <sub>eq</sub> (Å <sup>2</sup> )
U1	8h	0.09603(6)	0.59603(6)	0	0.0137(3)
U2	16k	0.10622(6)	0.21545(6)	0	0.0130(2)
Ni	4a	0	0	1/4	0.0136(8)

(b) Atomic position parameters for UNi<sub>5</sub>

Atom	Site	x	y	z	U <sub>eq</sub> (Å <sup>2</sup> )
U	4a	0	0	0	0.0137(3)
Ni1	16e	0.62385(2)	0.62385(2)	0.62385(2)	0.0168(4)
Ni2	4c	1/4	1/4	1/4	0.0244(7)

(c) Atomic position parameters for UNi<sub>2</sub>

Atom	Site	x	y	z	U <sub>eq</sub> (Å <sup>2</sup> )
U	4f	1/3	2/3	0.55813(7)	0.0193(4)
Ni1	2a	0	0	0	0.0130(2)
Ni2	6h	0.1658(2)	0.3316(4)	1/4	0.0156(5)

(d) Atomic position parameters for UNi<sub>1.7</sub>Si<sub>0.3</sub>

Atom	Site	x	y	z	U <sub>eq</sub> (Å <sup>2</sup> )
U	4f	1/3	2/3	0.5699(1)	0.0138(4)
M1 <sup>a</sup>	2a	0	0	0	0.0130(2)
Ni2	6h	0.1679(3)	0.3358(6)	1/4	0.0121(5)

<sup>a</sup> M1 = 39%Ni+61%Si

**Table 3**

(a) Main interatomic distances (Å) in U<sub>6</sub>Ni

U1 - 1U1	2.821(3)	U2 - 1U2	2.619(2)	Ni - 2Ni - 8U2	2.579(9) 2.808(6)
- 2U2	2.872(7)	- 2Ni	2.808(6)		
- 4U2	3.240(6)	- 1U1	2.872(7)		
4U1	3.260(7)	- 2U2	3.037(7)		

(b) Main interatomic distances (Å) in UNi<sub>5</sub>

U - 12Ni1	2.820(1)	Ni1 - 3Ni1	2.381(3)
- 4Ni2	2.943(1)	- 3Ni1	2.425(3)
- 12U	4.805(1)	- 3Ni2	2.815(1)
		- 3U	2.820(1)
Ni2 - 12Ni1	2.815(1)		
- 4U	2.943(1)		

(c) Main interatomic distances (Å) in UNi<sub>2</sub> (M1 = Ni) and UNi<sub>1.7</sub>Si<sub>0.3</sub> (M1 = 39%Ni+61%Si)

	UNi <sub>2</sub>	UNi <sub>1.7</sub> Si <sub>0.3</sub>
U - 3M1	2.909(1)	3.012(1)
- 3Ni2	2.924(1)	2.902(2)
- 6Ni2	2.946(1)	2.927(1)
- 3U	3.026(1)	3.158(1)
- 1U	3.167(1)	2.817(2)
M1 - 6Ni2	2.509(1)	2.459(2)
- 6U	2.909(1)	3.012(1)
Ni2 - 2Ni2	2.472(3)	2.547(5)
- 2Ni2	2.498(3)	2.584(5)
- 2M1	2.509(1)	2.459(2)
- 2U	2.924(1)	2.902(2)
- 4U	2.947(1)	2.927(1)

A complete investigation of the isothermal section at  $T = 900^\circ\text{C}$  of the U-Ni-Si ternary system [11] revealed that Si may substitute for Ni in UNi<sub>2</sub>, forming the solid solution UNi<sub>2-x</sub>Si<sub>x</sub> up to  $x = 0.3$ , keeping the MgZn<sub>2</sub> type. Refinement of the crystal structure of a single crystal having the limiting composition UNi<sub>1.7</sub>Si<sub>0.3</sub> and lattice parameters  $a = 5.1312(3) \text{ \AA}$ ,  $c = 7.8207(6) \text{ \AA}$  (Table 1) showed that Si replaces 61% of the Ni1 atoms on the 2a position. This substitution induces an increase of the U-M1 distance from 2.909(1)  $\text{ \AA}$  to 3.012(1)  $\text{ \AA}$  and a concomitant change of the shortest U-U distances, from 3.026(1)  $\text{ \AA}$  to 3.158(1)  $\text{ \AA}$  and from 3.167(1)  $\text{ \AA}$  to 2.817(2)  $\text{ \AA}$  (Table 2). Magnetic measurements for UNi<sub>1.7</sub>Si<sub>0.3</sub> indicated a nearly temperature-independent paramagnetism from 2 to 300 K, with a value of the magnetic susceptibility of  $9 \cdot 10^{-4} \text{ emu/mole}$  at 300 K. Consequently, the ferromagnetic correlations are strongly dependant on the modification of the band structure induced by the changes of the U-U distances from UNi<sub>2</sub> to UNi<sub>1.7</sub>Si<sub>0.3</sub>.

## References

- [1] A. Perricone, H. Noël, *J. Nucl. Mater.* 299 (2001) 260.
- [2] A. Perricone, M. Potel, H. Noël, *J. Alloys Compd.* 340 (2002) 39.
- [3] A. Perricone, H. Noël, *Intermetallics* 10 (2002) 519.
- [4] T.B. Massalski (Ed.), *Binary Alloy Phase Diagrams*, ASM, Materials Park, OH, USA, 1991, Vol. 3.
- [5] *Nonius Kappa CCD Program Package*, Nonius Delft, The Netherlands, 1998.
- [6] G.M. Sheldrix, *SHELX-97, Program for Crystal Structure Refinement*, University of Göttingen, Germany, 1997.
- [7] L.E. DeLong, R.P. Guertin, S. Hasanain, T. Fariss, *Phys. Rev. B* 31(11) (1985) 7059.
- [8] L. Havela, J. Hrebik, J. Sternberk, A. Menovsky, A. Zentko, *Phys. Status Solidi* 59 (1980) k165.
- [9] P.H. Frings, J.J.M. Franse, A. Menovsky, S. Zemirli, B. Barbara, *J. Magn. Magn. Mater.* 54 (1986) 541.
- [10] J.M. Fournier, A. Boeuf, P. Frings, M. Bonnet, J.X. Boucherle, A. Delapalme, A. Menovsky, *J. Less-Common Met.* 121 (1986) 249.
- [11] A. Perricone, *PhD thesis*, University of Rennes, 2002.

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