# Magnetic properties of solid solutions with $AlB_2$ -type structure in R-Ag-Al-Ge systems (R = Ce, Pr, Nd)

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For the AlB<sub>2</sub>-type solid solutions formed in the quaternary systems *R*-Ag-Al-Ge (R = Ce, Pr, Nd) magnetic measurements have been performed in the magnetic fields up to 5 T and in the temperature range 1.72-400 K. It is shown that these compounds fit to a Curie-Weiss expression above 80 K. The magnetic characteristics of the compounds (ordering temperature, Curie-Weiss temperature, effective magnetic moment, high-field magnetic moment) have been determined. The effective paramagnetic moments are close to the values of free  $R^{3+}$  (R = Ce, Pr, Nd) ions.

### Rare-earth / Silver / Aluminum / Germanium / Solid solution / X-ray diffraction / Magnetic measurements

## Introduction

Rare-earth alloys are attracting interest because they form a number of magnetic materials. High electrical and thermal conductivity, and good corrosion resistance are arguments for silver and its alloys to be used; corrosion resistance and lightness are typical features of aluminum alloys. Germanides of rare-earth elements have high melting points, high-temperature strength, special electrical and mechanical properties.

Several crystal structure investigations of intermetallics formed in R-Ag-Ge and R-Al-Ge systems (where *R* is a light rare-earth metal) have been carried out [1-7], but information about the magnetic properties is scarce. In particular, among the 12 ternary compounds with AlB<sub>2</sub>-type structure (Pearson space group code hP3.*P6/mmm*): (773 K), (773 K), LaAg<sub>0.8</sub>Ge<sub>1.2</sub> CeAg<sub>0.8</sub>Ge<sub>1.2</sub> CeAg<sub>0.8-0.67</sub>Ge<sub>1.2-1.33</sub> (1123 K), PrAg<sub>0.8</sub>Ge<sub>1.2</sub> (873 K), (1073 K),  $NdAg_{0.7}Ge_{1.3}$ EuAgGe (673 K), EuAg<sub>0.8</sub>Ge<sub>1.2</sub> (973 K), GdAg<sub>0.6</sub>Ge<sub>1.4</sub> (873 K) [1-3] and LaAl<sub>1.8-1.5</sub>Ge<sub>0.2-0.5</sub> (773 K), CeAl<sub>1.6-1.5</sub>Ge<sub>0.4-0.5</sub> (773 K), PrAl<sub>1.55-1.48</sub>Ge<sub>0.45-0.52</sub> (673 and 873 K), NdAl<sub>1.63-1.50</sub>Ge<sub>0.37-0.50</sub> (1273 K), SmAl<sub>1.5</sub>Ge<sub>0.5</sub> (1273 K), and EuAl<sub>1.08-1.00</sub>Ge<sub>0.92-1.00</sub> (873 K) [4-7], the magnetic behavior has been studied only for two of them, namely, CeAg<sub>0.67</sub>Ge<sub>1.33</sub> [8] and CeAl<sub>1.5</sub>Ge<sub>0.5</sub> [9]. According to [8] the  $CeAg_{0.67}Ge_{1.33}$  compound shows soft ferromagnetic properties with a Curie temperature of  $T_{\rm C} = 7.5$  K. The CeAl<sub>1.5</sub>Ge<sub>0.5</sub> compound orders ferromagnetically below  $T_{\rm C} = 6.15$  K [9].

The existence in the quaternary systems R-Ag-Al-Ge (R = Ce, Pr, Nd) at 873 K of complete solid solutions based on the ternary germanides CeAg<sub>0.8</sub>Ge<sub>1.2</sub>-CeAl<sub>1.6-1.5</sub>Ge<sub>0.4-0.5</sub>,  $PrAg_{0.8}Ge_{1.2}-$ PrAl<sub>1.55-1.48</sub>Ge<sub>0.45-0.52</sub>, and NdAg<sub>0.7</sub>Ge<sub>1.3</sub>-NdAl<sub>1.63-1.50</sub>Ge<sub>0.37-0.50</sub> with AlB<sub>2</sub>-type structure, was reported in [10]. The aim of this work was to investigate the magnetic properties of the AlB<sub>2</sub>-type solid solutions in the quaternary systems {Ce, Pr, Nd}-Ag-Al-Ge.

#### **Experimental**

17 four-component alloys with nominal compositions  $R_{33,3}(Ag,Al,Ge)_{66,7}$  were prepared by direct melting of the constituent elements (Ce, Pr, Nd  $\geq$  99.83 wt.%, Al  $\geq$  99.985 wt.%, Ag and Ge  $\geq$  99.999 wt.%) in an arc-furnace under purified (Ti-gettered) argon atmosphere on a water-cooled copper hearth. The samples were annealed at 873 K in evacuated quartz ampoules for 1 month and subsequently quenched in cold water. The weight loss during the preparation of the samples was less than 1% of the total mass, which was 1 g for each alloy.

Phase analysis was carried out using X-ray powder diffraction data collected on a diffractometer DRON-2.0M (Fe K $\alpha$  radiation) in the angular range  $20 \le 2\theta \le 100^{\circ}$ . The profile and structural parameters were refined by the Rietveld method, using the program DBWS-9807 [11]. Magnetization and dc

magnetic susceptibility measurements were performed on a SQUID magnetometer (Quantum Design MPMS) in magnetic fields up to 5 T and in the temperature range 1.72-400 K.

# **Results and discussion**

Structural (phases, their structure types, refined unitcell parameters) and magnetic (type of spin order, ordering and Curie-Weiss temperatures  $T_{\text{ord}}$  and  $\theta_{\text{p}}$ , effective magnetic moment  $\mu_{eff}$  and magnetic moment  $\sigma_{1.72K,5T}$  measured at T = 1.72 K in  $\mu_0$ H = 5 T) data for the samples of the *R*-Ag-Al-Ge systems (*R* = Ce, Pr, Nd) are listed in Tables 1-3. The temperature dependencies of the inverse magnetic susceptibility for the alloys at 0.1 T are shown it Figs. 1-3, where the solid line is a fit of the experimental data to the Curie-Weiss law. The insets display the field dependence of the magnetization at *T* = 1.72 K and its temperature dependence at 0.1 T.

**Table 1** Structural and magnetic data for the CeAg<sub>0.8</sub>Ge<sub>1.2</sub>–CeAl<sub>1.6-1.5</sub>Ge<sub>0.4-0.5</sub> solid solution.

Sample composition,	Dhasa	Structure	Unit	-cell param	Spin	$T_{\rm ord}$ ,	$\sigma_{ m s}$ ,	$\mu_{\rm eff}$ ,	
at.%	rnase	type	<i>a</i> , nm	<i>c</i> , nm	<i>V</i> , nm3	order K		$\mu_{\rm B}$	$\nu_p, \kappa_{\mu_B}$
Ce <sub>33.3</sub> Ag <sub>23.3</sub> Al <sub>6.7</sub> Ge <sub>36.7</sub>	CeAg <sub>0.70</sub> Al <sub>0.20</sub> Ge <sub>1.10</sub>	AlB <sub>2</sub>	0.43901(6)	0.41407(8)	0.06911(2)	F	5.5	0.9	-13.22.59
	CeAgGe	LiGaGe	0.4506(1)	0.7834(8)	0.1377(2)				
Ce33.3Ag20.0Al13.3Ge33.4	$CeAg_{0.60}Al_{0.40}Ge_{1.00}$	AlB <sub>2</sub>	0.43832(7)	0.41603(9)	0.06922(2)	F	5.7	1.2	12.2 2.58
Ce33.3Ag16.7Al20.0Ge30.0	CeAg <sub>0.50</sub> Al <sub>0.60</sub> Ge <sub>0.90</sub>	AlB <sub>2</sub>	0.43619(6)	0.42133(8)	0.06942(2)	F	3.3	0.7	-50.42.54
Ce33.3Ag13.3Al26.7Ge26.7	CeAg <sub>0.40</sub> Al <sub>0.80</sub> Ge <sub>0.80</sub>	AlB <sub>2</sub>	0.43487(6)	0.42474(7)	0.06956(2)	F	5.8	0.8	-17.32.50
Ce <sub>33.3</sub> Ag <sub>2.7</sub> Al <sub>46.7</sub> Ge <sub>17.3</sub>	CeAg <sub>0.08</sub> Al <sub>1.40</sub> Ge <sub>0.52</sub>	AlB <sub>2</sub>	0.43480(5)	0.42953(7)	$0.070\overline{32(2)}$	F	5.4	0.8	-19.02.44

Table 2 Structural and magnetic data for the PrAg<sub>0.8</sub>Ge<sub>1.2</sub>-PrAl<sub>1.55-1.48</sub>Ge<sub>0.45-0.52</sub> solid solution.

Sample composition,	Phase	Structure	Unit	Spin	$T_{\rm ord}$ ,	$\sigma_{ m s},$	$ heta_{\mathrm{p}},$	$\mu_{\rm eff}$ ,		
at.%		type	<i>a</i> , nm	<i>c</i> , nm	V, nm <sup>3</sup>	order	K	$\mu_{\rm B}$	K	$\mu_{\rm B}$
Pr33.3Ag23.3Al6.7Ge36.7	$PrAg_{0.70}Al_{0.20}Ge_{1.10}$	$AlB_2$	0.43722(6)	0.40981(7)	0.06785(2)	F	8.1	1.3	4.7	3.57
Pr33.3Ag16.7Al16.7Ge33.3	$PrAg_{0.50}Al_{0.50}Ge_{1.00}$	$AlB_2$	0.43415(6)	0.41652(8)	0.06799(2)	F	7.6	1.2	-0.6	3.65
Pr <sub>33.3</sub> Ag <sub>13.3</sub> Al <sub>20.0</sub> Ge <sub>33.4</sub>	PrAg <sub>0.40</sub> Al <sub>0.60</sub> Ge <sub>1.00</sub>	AlB <sub>2</sub>	0.43215(8)	0.41858(9)	0.06769(3)	F	7.6	1.2	-7.2	3.63
						F	42.5			
Pr <sub>33.3</sub> Ag <sub>12.7</sub> Al <sub>26.7</sub> Ge <sub>27.3</sub>	PrAg <sub>0.38</sub> Al <sub>0.80</sub> Ge <sub>0.82</sub>	$AlB_2$	0.43368(5)	0.41929(7)	0.06830(2)	F	7.6	1.4	2.9	3.62
Pr <sub>33.3</sub> Ag <sub>8.4</sub> Al <sub>33.3</sub> Ge <sub>25.0</sub>	PrAg <sub>0.25</sub> Al <sub>1.00</sub> Ge <sub>0.75</sub>	AlB <sub>2</sub>	0.43253(6)	0.42244(7)	0.06844(2)	F	3.2	1.3	-24.1	3.94
						AF	54.8			
Pr <sub>33.3</sub> Ag <sub>3.3</sub> Al <sub>43.4</sub> Ge <sub>20.0</sub>	$PrAg_{0.10}Al_{1.30}Ge_{0.60}$	AlB <sub>2</sub>	0.43288(9)	0.42389(9)	0.06880(2)	F	4.7	1.4	5.1	3.60

 $Table \ 3 \ Structural \ and \ magnetic \ data \ for \ the \ NdAg_{0.7}Ge_{1.3}-NdAl_{1.63-1.50}Ge_{0.37-0.50} \ solid \ solution.$ 

Sample composition,	Dhaga	Structure	Unit-ce	ell paramete	Spin	$T_{\rm ord}$ ,	$\sigma_{ m s},$	$\theta_{\mathrm{p}},$	$\mu_{\rm eff}$ ,	
at.%	rnase	type	<i>a</i> , nm	<i>c</i> , nm	V, nm <sup>3</sup>	order	K	$\mu_{\rm B}$	Ŕ	$\mu_{\rm B}$
Nd <sub>33.3</sub> Ag <sub>20.0</sub> Al <sub>6.7</sub> Ge <sub>40.0</sub>	NdAg <sub>0.60</sub> Al <sub>0.20</sub> Ge <sub>1.20</sub>	AlB <sub>2</sub>	0.43121(7)	0.41274(9)	0.06646(2)	F	4.9	1.9	0.7	3.68
						AF	10.5			
Nd <sub>33.3</sub> Ag <sub>17.3</sub> Al <sub>13.3</sub> Ge <sub>36.0</sub>	NdAg <sub>0.52</sub> Al <sub>0.40</sub> Ge <sub>1.08</sub>	AlB <sub>2</sub>	0.43379(6)	0.41144(8)	0.06705(2)	F	5.0	1.8	5.2	3.64
						AF	13.7			
$Nd_{33.3}Ag_{14.0}Al_{20.0}Ge_{32.7}$	NdAg <sub>0.42</sub> Al <sub>0.60</sub> Ge <sub>0.98</sub>	AlB <sub>2</sub>	0.43177(5)	0.41493(6)	0.06700(1)	F	5.1	1.8	5.3	3.60
						AF	11.9			
Nd <sub>33.3</sub> Ag <sub>11.3</sub> Al <sub>26.7</sub> Ge <sub>28.7</sub>	NdAg <sub>0.34</sub> Al <sub>0.80</sub> Ge <sub>0.86</sub>	AlB <sub>2</sub>	0.43048(5)	0.41811(6)	0.06710(1)	F	9.9	1.7	5.3	3.73
Nd <sub>33.3</sub> Ag <sub>6.7</sub> Al <sub>36.7</sub> Ge <sub>23.3</sub>	NdAg <sub>0.20</sub> Al <sub>1.10</sub> Ge <sub>0.70</sub>	AlB <sub>2</sub>	0.43138(4)	0.41951(5)	0.06761(1)	AF	2.5	1.7	1.0	3.73
Nd <sub>33.3</sub> Ag <sub>2.7</sub> Al <sub>46.7</sub> Ge <sub>17.3</sub>		AlB <sub>2</sub>	0.43138(5)	0.42100(6)	0.06785(1)	F	8.1	1.5	12.7	3.67
	NdAg <sub>0.08</sub> Al <sub>1.40</sub> Ge <sub>0.52</sub>					F	26.8			
						F	49.5			



**Fig. 1** Inverse magnetic susceptibility and magnetization as a function of temperature at 0.1 T and field dependence of the magnetization at T = 1.72 K for certain compositions of the solid solution  $CeAg_{0.8}Ge_{1.2}$ -CeAl<sub>1.6-1.5</sub>Ge<sub>0.4-0.5</sub>.



**Fig. 2** Inverse magnetic susceptibility and magnetization as a function of temperature at 0.1 T and field dependence of the magnetization at T = 1.72 K for the certain compositions of the solid solutions  $PrAg_{0.8}Ge_{1.2}-PrAl_{1.55-1.48}Ge_{0.45-0.52}$ .



**Fig. 3** Inverse magnetic susceptibility and magnetization as a function of temperature at 0.1 T and field dependence of the magnetization at T = 1.72 K for the certain compositions of the solid solutions  $NdAg_{0.7}Ge_{1.3}-NdAl_{1.63-1.50}Ge_{0.37-0.50}$ .

According to the X-ray powder diffraction patterns all of the synthesized alloys are single-phase samples (except Ce<sub>33.3</sub>Ag<sub>23.3</sub>Al<sub>6.7</sub>Ge<sub>36.7</sub>) and their crystal structures belong to the hexagonal AlB<sub>2</sub> type. Within the homogeneity ranges CeAg<sub>0.8</sub>Ge<sub>1.2</sub>-CeAl<sub>1.6-1.5</sub>Ge<sub>0.4-0.5</sub>, PrAg<sub>0.8</sub>Ge<sub>1.2</sub>-PrAl<sub>1.55-1.48</sub>Ge<sub>0.45-0.52</sub>, and  $NdAg_{0.7}Ge_{1.3}-NdAl_{1.63-1.50}Ge_{0.37-0.50}$ , the a-parameter decreases with increasing Al content (0-54.3 at.%) and decreasing Ag and Ge content (26.7-0 and 43.3-12.3 at.%, respectively), whereas the c-parameter increases. The replacement of Ag and Ge atoms (covalent radii r = 0.134 and 0.122 nm, respectively) by Al atoms (r = 0.118 nm) influences mainly the *a*-parameter, which reflects the contact distances between the small atoms in the structure  $(\delta_{M-M} = a/\sqrt{3})$ . The variation of the unit-cell parameters within the homogeneity ranges of the AlB<sub>2</sub>-type compounds in the quaternary systems {Ce, Pr, Nd}–Ag–Al–Ge is shown in Fig. 4.

The magnetic measurements show that the magnetic susceptibility of the  $AlB_2$ -type solid solutions follows the Curie-Weiss law

$$\chi_{\rm M} = \frac{\rm C_{\rm M}}{T - \theta_{\rm p}}$$

in the temperature range 20-400 K for CeAg<sub>0.8</sub>Ge<sub>1.2</sub>-CeAl<sub>1.6-1.5</sub>Ge<sub>0.4-0.5</sub>, 80-400 K for PrAg<sub>0.8</sub>Ge<sub>1.2</sub>-PrA11.55-1.48Ge0.45-0.52 and 40-400 K for NdAg0.7Ge1.3-NdAl<sub>1.63-1.50</sub>Ge<sub>0.37-0.50</sub>. As a rule, at low temperatures fairly complex magnetic ordering with notable ferromagnetic component is observed for these compounds. For the solid solution NdAg<sub>0.7</sub>Ge<sub>1.3</sub>-NdAl<sub>1.63-1.50</sub>Ge<sub>0.37-0.50</sub> the replacement of Ag and Ge by Al atoms leads to an increase of the ordering temperature  $T_{\rm C}$  and a decrease of the moment  $\sigma_{1.72\rm K,5T}$ . For the solid solution PrAg<sub>0.8</sub>Ge<sub>1.2</sub>-PrAl<sub>1.55-1.48</sub>Ge<sub>0.45-0.52</sub> the replacement of Ag and Ge by Al atoms results in a decrease of  $T_{ord}$ , while  $\sigma_{1.72K,5T}$  remains almost constant. For the investigated compositions of the solid solution CeAg<sub>0.8</sub>Ge<sub>1.2</sub>– CeAl<sub>1.6-1.5</sub>Ge<sub>0.4-0.5</sub>,  $T_{ord}$  is ~5.6 K and  $\sigma_{1.72K,5T}$  decreases slightly when Ag and Ge atoms are replaced by Al atoms. The small hysteresis effect observed for the alloys indicates soft character of their magnetism. The measured effective paramagnetic moments per formula unit, calculated by the formula

$$\mu_{\rm eff} = 2.828 \sqrt{\chi_{\rm M} \cdot (T_i - \theta_{\rm p})} \,,$$

are close to the values of free  $R^{3+}$  ions (the theoretical effective paramagnetic moments are 2.54 for Ce<sup>3+</sup>, 3.58 for Pr<sup>3+</sup>, and 3.62 for Nd<sup>3+</sup>).

## Conclusions

An investigation of the magnetic properties of the solutions CeAg<sub>0.8</sub>Ge<sub>1.2</sub>-CeAl<sub>1.6-1.5</sub>Ge<sub>0.4-0.5</sub>, solid PrAg<sub>0.8</sub>Ge<sub>1.2</sub> - PrAl<sub>1.55-1.48</sub>Ge<sub>0.45-0.52</sub>, and NdAg<sub>0.7</sub>Ge<sub>1.3</sub>-NdAl<sub>1.63-1.50</sub>Ge<sub>0.37-0.50</sub> with AlB<sub>2</sub>-type structure has been carried out. The results of the dc magnetic susceptibility measurements revealed the Curie-Weiss paramagnetic behavior. The effective paramagnetic moments of the compounds measured above 80 K are close to the respective values of the free  $R^{3+}$ (R = Ce, Pr, Nd) ions, which is an evidence of the trivalent state of the rare-earth atoms. These results indicate that only the rare-earth atoms carry magnetic moments. At low temperatures, all the compounds studied order magnetically. The observed fairly complex behaviors of the magnetization as a function of temperature and magnetic field strength hint at complicated magnetic structures and multiple phase transitions occurring in most of these materials.



**Fig. 4** Unit-cell parameters within the homogeneity ranges of the  $AlB_2$ -type compounds in the quaternary systems {Ce,Pr,Nd}–Ag–Al–Ge as a function of the valence electron concentration VEC<sub>A</sub>.

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